

## Lecture 12: Mechanism of Oxygen carriage in the blood

Code: RRS-209

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### Learning Objective :

#### Knowledge:

- Know how the oxygen is carried by the blood.
- Define  $O_2$  content,  $O_2$  saturation,  $O_2$  utilization.
- Describe the  $O_2$  dissociation curve and the physiological significance of the curve.
- Know the causes of failure of oxygen transport and cyanosis.
- Explain myoglobin dissociation curve, fetal Hbg.

#### Intellectual:

- Explore the factors causes shifting of  $O_2$  dissociation curve.
- Compare between types of hypoxia

### ***O<sub>2</sub> Carriage by the blood***

- Oxygen is transported in the blood from the alveolar capillaries of the lungs (where blood is loaded with O<sub>2</sub>) to the peripheral capillaries in the tissues.
- O<sub>2</sub> is transported in blood in two distinct ways:

1- Bound to hemoglobin (Hb) **OR**

2- Dissolved in solution in ICF and ECF fluids

#### **O<sub>2</sub> in Physical solution:**

- The amount of oxygen dissolved in the blood is proportional to its partial pressure (Henry's law).
- At 37°C, 3ml O<sub>2</sub> is dissolved in each liter of arterial blood per mmHg. So there is 3 ml / liter. In whole blood volume = 3X5= 15 ml.
- Resting O<sub>2</sub> consumption is approximately (300L /min) So the physical form of O<sub>2</sub> can not support the body's O<sub>2</sub> requirement.
- However, dissolved O<sub>2</sub> determine the major pathway (direction of diffusion of O<sub>2</sub>) for transport of O<sub>2</sub> across capillary walls to the cells.
- So an additional form of O<sub>2</sub> transport is needed. Hemoglobin provides this transport.

#### **2- Chemical combination with Hb (98.5%)**

- Hb contains 4 atoms of iron. Each atom combines with one molecule of O<sub>2</sub> = Hb<sub>4</sub>O<sub>8</sub> ( *there are only four ' hooks ' for O<sub>2</sub> per molecule* )
- **O<sub>2</sub> content** : The amount of O<sub>2</sub> that can be bound to hemoglobin (mL/dL blood) in a liter of arterial blood.
- calculated as 1.34 mL O<sub>2</sub>/dL blood × [Hemoglobin].
- As 1gm Hbg contains 1.34 ml O<sub>2</sub> and every 100 ml blood contain 15 gram of Hb.
- in arterial blood = 1.34 X 15 = 20.1ml O<sub>2</sub>/100 ml blood Venous blood contains 15 ml/100 ml blood.
- So 1L of blood containing 150g Hb can transport 200 ml. Compare this value with that of dissolved form
- **O<sub>2</sub> saturation**: The percentage of total oxygen-binding sites on hemoglobin that are actually occupied by oxygen, also called the saturation of peripheral oxygen.
- **O<sub>2</sub> utilization**: Every 100 ml of arterial blood while passing in the tissues loses O<sub>2</sub> and changes to venous blood.
- Every 100 ml of arterial blood loses 5ml (20-15gm) = (50 ml / liter) to the tissues.

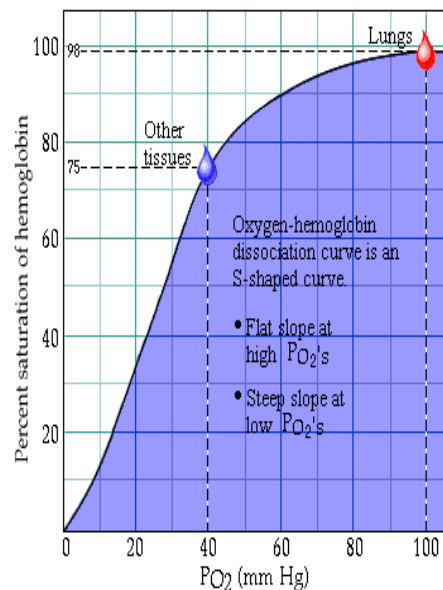
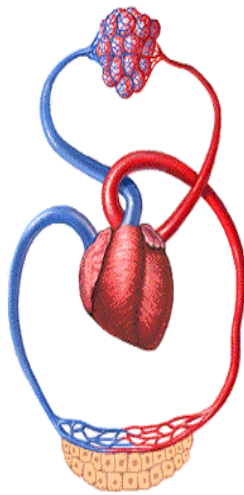
$$\text{Coefficient O}_2 \text{ Utilization} = \frac{\text{Arterial O}_2 \text{ content} - \text{venous O}_2 \text{ content}}{\text{Arterial O}_2 \text{ content}} \times 100$$

## O<sub>2</sub> Dissociation Curve

- It's the relationship between the O<sub>2</sub> tension (PO<sub>2</sub>) and % HbO<sub>2</sub> saturation. It describes how the oxygen saturation of hemoglobin varies with the Po<sub>2</sub> in the blood
- The reaction between Hb and O<sub>2</sub> is both rapid and reversible.
- It's not linear but it's S (sigmoid) shaped curve. The curve has a steep slope between 10 and 60 mmHg PO<sub>2</sub> and a relatively flat portion (or plateau) between 60 and 100 mmHg PO<sub>2</sub>.

### Causes of the sigmoid shape:

- The intermediates compounds (Hb<sub>4</sub>O<sub>4</sub> & Hb<sub>4</sub>O<sub>6</sub>) are responsible for the S shaped. If these compounds aren't formed and Hb<sub>4</sub>O<sub>8</sub> is formed directly the curve would be straight line.
- Binding of O<sub>2</sub> to Hb is cooperative such that the binding of each O<sub>2</sub> molecule to the Hb tetramer facilitates the binding of the next
- So, the combination of the 1<sup>st</sup> heam with O<sub>2</sub> ↑ the affinity of the 2<sup>nd</sup> heam for O<sub>2</sub> and oxygenation of the 2<sup>nd</sup> ↑ affinity of the 3<sup>rd</sup> heam for O<sub>2</sub> and so on.
- $\text{Hb}_4 + \text{O}_2 \rightarrow \text{Hb}_4\text{O}_2$
- $\text{Hb}_4\text{O}_2 + \text{O}_2 \rightarrow \text{Hb}_4\text{O}_4$
- $\text{Hb}_4\text{O}_4 + \text{O}_2 \rightarrow \text{Hb}_4\text{O}_6$
- $\text{Hb}_4\text{O}_6 + \text{O}_2 \rightarrow \text{Hb}_4\text{O}_8$



- The major function of Hb is to *load with* O<sub>2</sub> at the lungs and *unload at the tissues*. This function is carried out at the flat (**loading region**) part of the curve and at the steep **unloading region**.

**The physiological significance of the flat part (used at the lungs):**

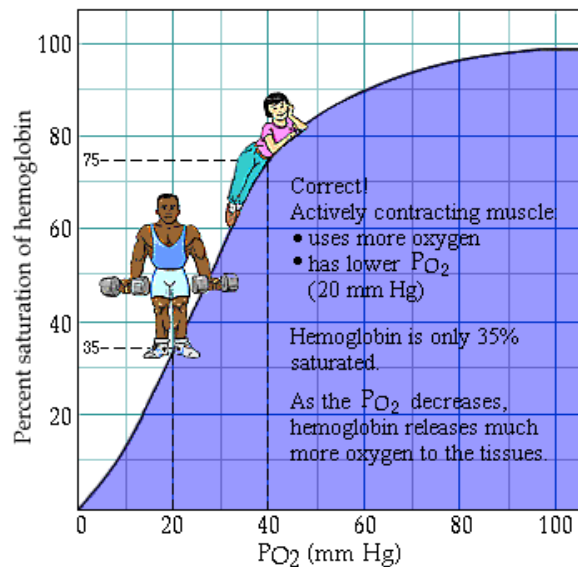
- At 100 mmHg O<sub>2</sub> tension, the Hb is 98 % saturated.
- At 80 mmHg O<sub>2</sub> tension, the Hb is 95 %saturated.
- At 60 mmHg.O<sub>2</sub> tension, the Hb is 90 % saturated.

**From the curve you can observe**

- ✓ Thus, despite the marked fall in alveolar Po<sub>2</sub> from 100 to 60 mm Hg, the Hbg saturation changed from (98%) to (90%) which is still within normal levels. Thus the tissue Po<sub>2</sub> hardly changes

**The physiological significance of the flat part (used at the lungs):**

- **Note :** Even a small fall in blood Po<sub>2</sub> causes a large unloading of O<sub>2</sub> .
- At O<sub>2</sub> tension = 40 mmHg, the Hb saturation is 75 % . So, Hb saturation decreases by **20 %** (95% - 75 %).
- Significance: During muscular exercise, PO<sub>2</sub> is ranged from 15 -30 mmHg. The Hb saturation is 35% . So, Hb saturation decreases by **60%** (3 times the normal) (95-35%). Which means that there is more O<sub>2</sub> delivered to tissues as they need more amount of O<sub>2</sub> due to ↑ the activity of them.



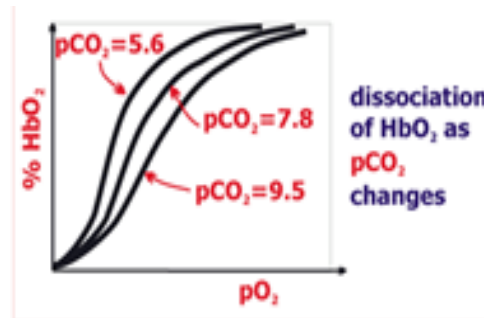
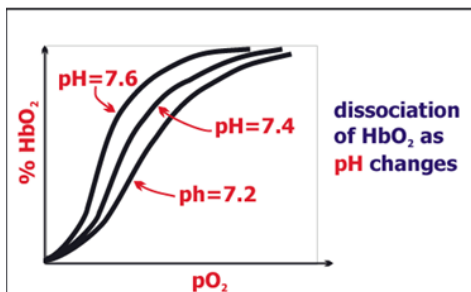
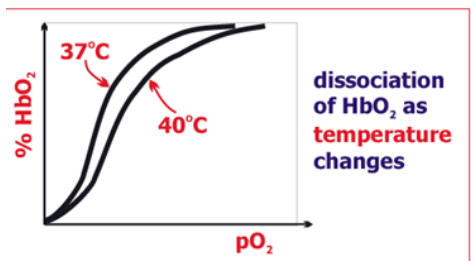
**Factors affecting  $O_2$  D curve:**

Factors shifting to the right: It means the affinity of Hb to  $O_2$  is decreased OR more release of  $O_2$  from the Hb, caused by:

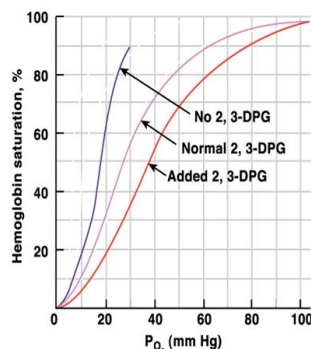
- 1- ↑ temperature
- 2- ↑  $CO_2$  concentration in the blood.
- 3- ↓ in PH of the blood = ↑ in  $H^+$  conc.
- 4- ↑ in concentration Of 2'3 Diphosphoglycerate ( DPG)

Shifting to the left: (The affinity to  $O_2$  is increased) OR more bind of  $O_2$

- 1- ↓ temperature
- 2- ↓  $CO_2$  conc. In the blood.
- 3- ↑ in PH of the blood = ↓ in  $H^+$  conc.
- 4- ↓ in conc. Of 2'3 DPG.

**4- Effect of 2,3 Diphosphoglycerate (2,3 DPG)**

- 2,3-DPG is produced by erythrocytes during glycolysis, binds to Hb and reduces its affinity for  $O_2$ . The production of 2, 3-DPG is raised during hypoxic conditions, the



**Graph question: Blood stored in blood banks loses its normal content of 2, 3-DPG. Is this good or bad? Explain.**

**Significance of shift O<sub>2</sub> D curve :****1-The Bohr Effect:**

It is the ↑ of O<sub>2</sub> delivery to the tissues when CO<sub>2</sub> and H<sup>+</sup> shift the curve to the right.

**In the lungs:**

- As the blood passes in the lungs, CO<sub>2</sub> diffuses (Why ?) from blood into the alveoli, this ↓ the blood PCO<sub>2</sub> so ↓ H<sup>+</sup> concentration.
- This leads to shift to left → More binding of O<sub>2</sub> to Hb so oxygenation of the blood occurs

**In tissues:**

- When blood reaches the tissues , the CO<sub>2</sub> diffuse from the tissues to the blood , so PCO<sub>2</sub>↑ so shift the curve to the right which cause more release of O<sub>2</sub> to the tissue.

**2- In muscular exercise :**

- There are high CO<sub>2</sub> amount released and acids are produced .In addition, temperature of the muscle rises from 2-3C. All these factors cause shift the O<sub>2</sub> hemoglobin dissociation curve to right which allows more release of O<sub>2</sub> to the muscle.

**Variant types of HB****1-Fetal haemoglobin (HbF) has a raised affinity for O<sub>2</sub>**

- Compared with adult haemoglobin. This allows an increase in oxygen uptake in the placenta. Therefore, although fetal arterial PO<sub>2</sub> is lower than that in the air-breathing newborn, fetal hemoglobin allows adequate oxygen supply to the developing organs.

**2- O<sub>2</sub> D curve of myoglobin:**

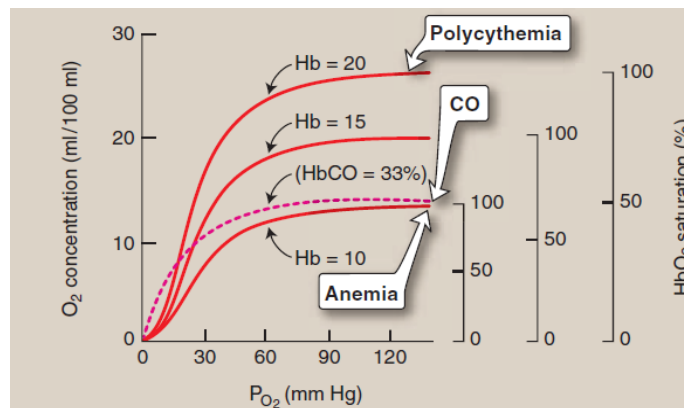
- Is a form of haemoglobin expressed in striated muscle fibers. It has a much higher affinity for O<sub>2</sub> than haemoglobin and does not demonstrate cooperativity in its binding of O<sub>2</sub>
- It can combine with one molecule of O<sub>2</sub> . and does not demonstrate cooperativity in its binding of O<sub>2</sub>
- The curve is rectangular hyperbola (remains horizontal till very low O<sub>2</sub> tension then suddenly descends vertically ).
- **So it acts as store of O<sub>2</sub>** to be used by muscle where O<sub>2</sub> tension becomes very low in tissues (as in severe exercise or hypoxic conditions), and also allows O<sub>2</sub> to be delivered to cells when muscle is contracted and perfusion reduced.

### 3- Carbon monoxide:

- It is a colorless, odorless gas that is a product of the incomplete combustion of fuel (e.g. gasoline).
- It is a common cause of sickness and death due to poisoning,
- It has extremely high affinity— 210 times that of oxygen—for the oxygen-binding sites in hemoglobin. For this reason, it reduces the amount of oxygen that combines with hemoglobin in pulmonary capillaries
- It also exerts a second deleterious effect as it **shift the oxygen-hemoglobin dissociation curve to the left**, thus decreasing the unloading of oxygen from hemoglobin in the tissues.
- Why Carbon monoxide is highly toxic gas?
- 1-The affinity of Hbg to CO is 210 times its affinity for O<sub>2</sub> . So, Once Hb combines with CO, it can not combine with O<sub>2</sub> . 2- The Hb CO shift the O dissociation curve of the remaining oxy Hbg to the left.
- The Hb CO breaks down very slowly.



Describe effects (Polycythemia, Anemia and CO on Hb concentration)



### Hypoxia (Failure of O<sub>2</sub> transport)

- The term for a lack of oxygen in the tissues is **hypoxia** . Lack of O<sub>2</sub> in arterial blood is termed **hypoxaemia** . Total absence of O<sub>2</sub> is **anoxia**.

#### ➤ Types of hypoxia :

- 1- Hypoxic hypoxia
- 3- Stagnant hypoxia

- 2- Anaemic hypoxia
- 4- Histotoxic hypoxia

**1- Hypoxic Hypoxia:**

- In this type , there is ↓ PO<sub>2</sub> of arterial blood, The Hbg saturation with O<sub>2</sub> is decreased and there is ↓ PO<sub>2</sub> of venous blood.
- When there is decrease in arterial PO<sub>2</sub> which goes to the tissues, there will be decrease in average PO<sub>2</sub> in capillary blood so the rate of O<sub>2</sub> diffusion to the tissues is ↓ which causes symptoms of O<sub>2</sub> lack.

**Causes:**

- 1- High altitude
- 2- Breathing low % of O<sub>2</sub>.
- 3- Shallow rapid breathing (*may results from pulmonary congestion*) because there is :
  - a- ↑ ratio of the volume of the DS to Tidal air.
  - b- Greater number of alveoli will not be distensible.
- 4- Depression of respiratory centers: as in morphine poisoning.
- 5-Diseases of the lung : may cause hypoxia but with different mechanisms:
  - a- By diffusion impairment : due to thickened pulmonary membrane as in pneumonia, pulmonary oedema
  - b- By decreasing surface area : emphysema.
  - c- Difficulty in breathing : Bronchial asthma in which there is increased resistance to air flow in the respiratory passages.
- 6- Shunting of venous blood.

**2- Anaemic Hypoxia**

- In this type; Normal Pa O<sub>2</sub> ,normal % saturation of Hb.
- Arterial O<sub>2</sub> content is ↓because of ↓ Hb amount which is capable of carrying O<sub>2</sub>.
- During passage of the blood in the tissues, a fewer number of RBCs. passes through the tissues and, so the O<sub>2</sub> tension decreased in the venous blood and then it decreased in the capillary blood leading to production of the hypoxic symptoms.

**Causes:**

- 1- All types of anemia.
- 2- Carbon monoxide (CO) poisoning.

**3- Stagnant Hypoxia**

In this type:

Normal Pa O<sub>2</sub>, normal % saturation of Hbg.

Caused by decreased blood flow through the tissues ,May be :

- 1- Generalized (congestive heart failure)
- 2- Localized (Cold).



**4- Histotoxic Hypoxia**

- O<sub>2</sub> released from Hbg is transported to the cell by the cytochrome system.
- Histotoxic Hypoxia results from inactivation of metabolic enzymes which facilitate this transport. These enzymes are Cytochrome dehydrogenase and cytochrome oxidase
- *Cyanide can block the cytochrome oxidase & Alcohol block the cytochrome dehydrogenase.*

***Cyanosis***

- Means the blue discoloration of the skin and mucous membrane, due to excessive amounts of deoxygenated Hbg in skin vessels.
- Threshold of cyanosis:
- Appears when the arterial blood contains more than 5 grams deoxygenated Hbg in each deciliter of blood.
- Causes:

1-Alveolar hypoventilation:

2-Inadequate oxygenation: as in deficiency of O in atmosphere.

3- Diffusion impairment

4- Ventilation-perfusion mismatch

4- Right to left shunt.

5- Circulatory defect (generalized & localized)

6- Abnormal forms of Hb

**Central cyanosis:**

- Caused by reduced O<sub>2</sub> saturation. Involves highly vascularized tissues such as lips and tongue and mucous membrane.

**Peripheral cyanosis:**

- Results from increased oxygen extraction from the peripheral blood resulting from sluggish movement of blood through capillary circulation.
- Affects distal extremities

Cyanosis occurs in moderate cold in exposed areas in normal individuals because there is arteriolar and venous constriction and there is slow blood flow in the capillaries & more oxygen is removed from Hg.

**Cyanosis is not present in:**

Sever cold: the drop in temperature causes shift to the left and the O<sub>2</sub> uptake of the cold tissues is reduced.